# Brevia

# H<sub>2</sub>S Induces a Suspended Animation–Like State in Mice

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Many organisms respond to changes in environmental conditions by entering into a suspended animation–like state in which a decrease in metabolic rate (MR) is followed by a reduction in core body temperature (CBT) (1). Regulated induction of a hypometabolic state is hypothesized to have great medical benefit for a variety of conditions, including ischemia and reperfusion injury, pyrexia, and other trauma (2). Suspended animation–like states may also be useful for creating beneficial hypothermia in surgical situations and for improving organ preservation (1).

Inhibiting oxidative phosphorylation reversibly induces states of profound hypometabolism in several model organisms (3–5). Because hydrogen sulfide (H<sub>2</sub>S) is a specific, potent, and reversible inhibitor of complex IV (cytochrome c oxidase), the terminal enzyme complex in the electron transport chain (6), we hypothesized that it could reduce MR and CBT in mammals.

When mice were exposed to 80 ppm of  $H_2S$ , their oxygen (O<sub>2</sub>) consumption dropped by ~50% and their carbon dioxide (CO<sub>2</sub>)

output dropped by  $\sim 60\%$  within the first 5 minutes (Fig. 1A) (7). If left in this environment for 6 hours, their MR dropped by  $\sim$ 90% (Fig. 1A). The MR of control mice, as judged from O<sub>2</sub> consumption and CO<sub>2</sub> output increases  $(\delta)$ . This drop in MR was followed by a drop in CBT to  $\sim 2^{\circ}$ C above ambient temperature (Fig. 1B). The average CBT of these mice reached a minimum of 15°C in an ambient temperature of 13°C (Fig. 1B). At this minimum CBT, both CO<sub>2</sub> output and  $O_2$  consumption was ~10% of normal (Fig. 1A), and the breathing rate of the mice decreased from  $\sim 120$  breaths per minute (BPM) to less than 10 BPM (8). After 6 hours of exposure to H<sub>2</sub>S, the mice were returned to room air and temperature, and their MR and CBT returned to normal (Fig. 1, A and B).

Exposing mice to varying concentrations of  $H_2S$  revealed a linear relationship between the concentration of  $H_2S$  and CBT (Fig. 1C). CBT dropped faster and reached lower temperatures as concentrations of  $H_2S$  increased from 0 to 80 ppm (8), suggesting that the effects of  $H_2S$  are concentration-dependent.



**Fig. 1.** CBT and MR of mice exposed to  $H_2S$ . (A) Relative  $CO_2$  production and  $O_2$  consumption of mice exposed to 80 ppm of  $H_2S$ . (B) CBT of mice during 6 hours of exposure to either 80 ppm of  $H_2S$  (black line) or the control atmosphere (gray line). The dotted line indicates ambient temperature. Values in (A) and (B) are means  $\pm$  one standard deviation. (C) Linear relationship between  $H_2S$  concentration and CBT ( $R^2 = 0.95$ ) after 6 hours of exposure. (D)  $CO_2$  output and CBT of mice (time = 0 at the start of  $H_2S$  exposure).

However, this MR reduction is not dependent on ambient temperature (fig. S1).

Because  $H_2S$  can be toxic in high doses, we conducted behavioral and functional tests, selected from the SHIRPA protocol (9), to assay for  $H_2S$ -induced damage. No behavioral or functional differences in the mice were detected after exposure to 80 ppm of  $H_2S$  for 6 hours (8). In the absence of  $H_2S$ , no effect on CBT was observed (Fig. 1B, control atmosphere). In addition, others report no long-term health effects with these  $H_2S$  concentrations (6).

The sequential drop in MR and CBT observed in mice (Fig. 1D) exposed to 80 ppm of  $H_2S$  is similar to that observed when animals initiate hibernation, daily torpor, or estivation (1). On-demand induction of a suspended animation–like state could provide insight into the mechanisms that govern natural states of reduced metabolism. Lowering metabolic demand in this way could be used to reduce physiological damage resulting from trauma and might improve outcomes after surgery.

#### References and Notes

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- Materials and methods are available as supporting online material on Science Online.
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### Supporting Online Material

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Materials and Methods SOM Text

Fig. S1

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